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## Key indicators

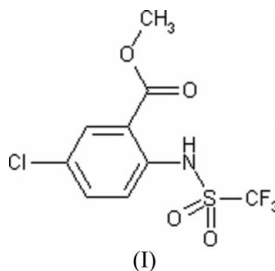
Single-crystal X-ray study  
 $T = 296$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.011$  Å  
 $R$  factor = 0.105  
 $wR$  factor = 0.284  
Data-to-parameter ratio = 11.7For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

## Methyl 5-chloro-2-[[trifluoromethyl)sulfonyl]amino]benzoate

The title compound [systematic name: methyl 5-chloro-2-((trifluoromethyl)sulfonamido)benzoate],  $\text{C}_9\text{H}_7\text{ClF}_3\text{NO}_4\text{S}$ , is a novel acaricide commonly named amidoflumet. The orientations of two side chains are such that the sulfonamide H atom and the carbonyl O atom of the ester substituent are coplanar with the aromatic ring, forming an intramolecular N—H...O hydrogen bond.

## Comment

The title compound, amidoflumet, (I) (Fig. 1), is a novel acaricide developed by Sumitomo Chemical Co. Ltd (Mori *et al.*, 2004). Compound (I) dissolves well in several organic solvents (*N,N*-dimethylformamide, chloroform, acetonitrile, methanol, ethanol *etc*) but is hardly soluble in water.



There is an intramolecular hydrogen bond (Table 2) between the sulfonamide H atom and the carbonyl O atom of the ester substituent. The crystal used was a little fragile and structural anisotropy was observed. These characteristics are a consequence of the lack of major intermolecular interactions along the *c* axis (Fig. 2).

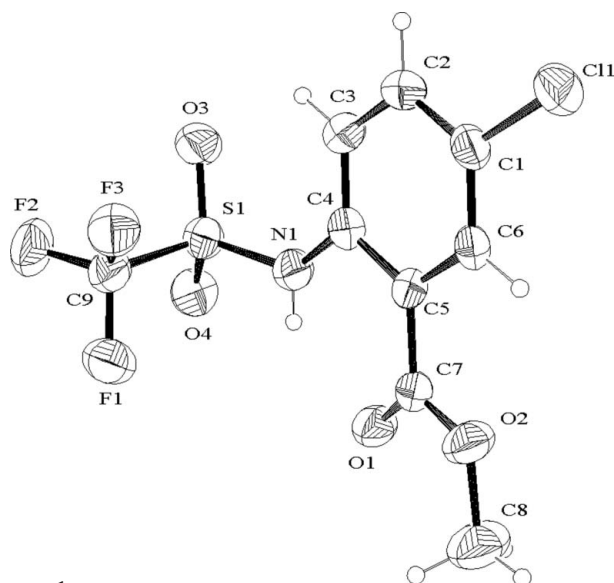
## Experimental

The analytical standard of (I) used for this study was manufactured and purified by Sumitomo Chemical Co. Ltd. The purity was determined by high-performance liquid chromatography and differential thermal analysis to be 100%. Crystallization was performed by gradual cooling of a supersaturated solution of (I) in chloroform at 300 K after mild heating to dissolve the residue completely. Elemental analysis calculated for  $\text{C}_9\text{H}_7\text{ClF}_3\text{NO}_4\text{S}$ : C 34.0, H 2.2, N 4.4, F 17.9, Cl 11.2, S 10.1%; found C 33.9, H 2.4, N 4.2, F 17.8, Cl 11.1, S 10.5%. IR (KBr,  $\text{cm}^{-1}$ ): 2900–3100 (*w*), 1695 (*ms*), 1490 (*ms*), 1310 (*ms*), 1200 (*s*), 1150 (*ms*);  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ , p.p.m.):  $\delta$  4.00 (3H, *s*), 7.54 (1H, *dd*), 7.72 (1H, *d*), 8.05 (1H, *d*), 11.2 (1H, *s*);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ , p.p.m.):  $\delta$  53.27, 117.68, 119.63 (splitting to four peaks because of C–F coupling), 120.58, 130.24, 130.95, 134.82, 136.96, 167.55; EI–MS fragmentation:  $m/z = 317, 285, 248, 216, 184, 154$ .

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**Figure 1**  
The molecular structure of (I), with the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probability level.

#### Crystal data

$C_9H_7ClF_3NO_4S$   
 $M_r = 317.67$   
 Triclinic,  $P\bar{1}$   
 $a = 5.341$  (2) Å  
 $b = 9.887$  (1) Å  
 $c = 11.754$  (3) Å  
 $\alpha = 85.212$  (9)°  
 $\beta = 85.47$  (1)°  
 $\gamma = 88.939$  (6)°  
 $V = 616.5$  (3) Å<sup>3</sup>

$Z = 2$   
 $D_x = 1.711$  Mg m<sup>-3</sup>  
 Cu  $K\alpha$  radiation  
 Cell parameters from 3856 reflections  
 $\theta = 3.8$ – $66.8$ °  
 $\mu = 4.84$  mm<sup>-1</sup>  
 $T = 296.1$  K  
 Block, colorless  
 $0.30 \times 0.20 \times 0.10$  mm

#### Data collection

Rigaku R-AXIS RAPID diffractometer  
 $\omega$  scans  
 Absorption correction: multi-scan (ABSCOR; Higashi, 1995)  
 $T_{\min} = 0.178$ ,  $T_{\max} = 0.616$   
 12598 measured reflections

2073 independent reflections  
 1398 reflections with  $F^2 > 2\sigma(F^2)$   
 $R_{\text{int}} = 0.127$   
 $\theta_{\max} = 68.2$ °  
 $h = -6 \rightarrow 6$   
 $k = -10 \rightarrow 10$   
 $l = -14 \rightarrow 14$

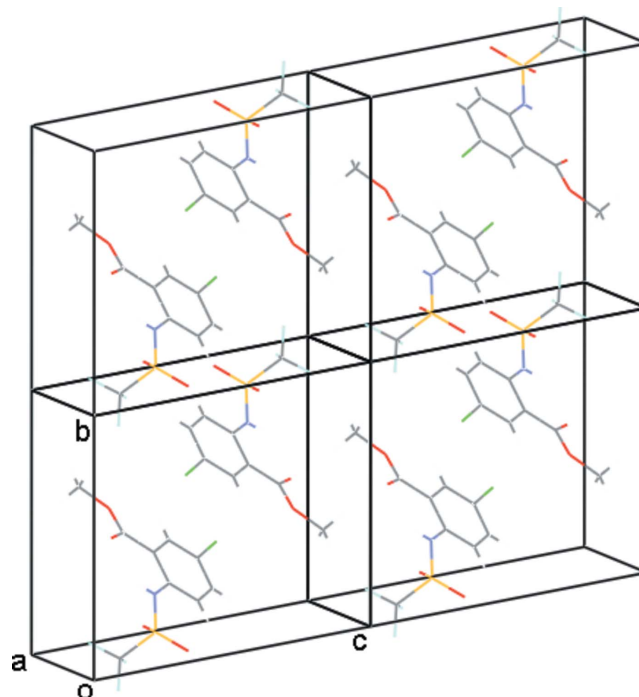
#### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.105$   
 $wR(F^2) = 0.284$   
 $S = 1.06$   
 2073 reflections  
 177 parameters  
 H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.1063P)^2 + 1.5441P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} < 0.001$   
 $\Delta\rho_{\max} = 0.36$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.35$  e Å<sup>-3</sup>

**Table 1**  
Selected geometric parameters (Å, °).

S1–N1	1.586 (8)	N1–C4	1.42 (1)
O4–S1–O3	122.6 (4)	S1–N1–C4	131.0 (6)
C9–S1–N1	104.1 (4)		
C9–S1–N1–C4	–79.3 (7)	C4–C5–C7–O1	9 (1)
S1–N1–C4–C5	158.8 (7)		



**Figure 2**  
The packing of (I).

**Table 2**  
Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N1–H7 $\cdots$ O1	0.78 (8)	2.01 (8)	2.606 (9)	132 (8)

The H atom bonded to the N atom was located in a difference map and refined isotropically. The H atoms bonded to C atoms were positioned geometrically ( $C-H = 0.95$  Å) and refined with riding-model constraints and with  $U_{\text{iso}}(H) = 1.2U_{\text{eq}}(C)$ .

Data collection: *PROCESS-AUTO* (Rigaku, 1998); cell refinement: *PROCESS-AUTO*; data reduction: *CrystalStructure* (Rigaku/MS & Rigaku, 2004); program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP III* (Burnett & Johnson, 1996); software used to prepare material for publication: *CrystalStructure*.

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